

### Preliminary communication

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## CRYSTAL AND MOLECULAR STRUCTURES OF *cis*-DICHLOROBIS(TRIFLUOROPHOSPHINE)BIS(TRIPHENYLPHOSPHINE)RUTHENIUM(II), *cis*-RuCl<sub>2</sub>(PF<sub>3</sub>)<sub>2</sub>(PPh<sub>3</sub>)<sub>2</sub>, AND DICHLORO(2,7-DIMETHYLOCTA-2,6-DIENE-1,8-DIYL)(TRIFLUOROPHOSPHINE)RUTHENIUM(IV), RuCl<sub>2</sub>(PF<sub>3</sub>)(C<sub>10</sub>H<sub>16</sub>)

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### Summary

The first structures of trifluorophosphine complexes of transition metals in an oxidation state greater than +1 are reported.

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Although many transition metal complexes of trifluorophosphine have been characterised [1-3], very little structural work on them has been reported, and in no case are there details on complexes in which the oxidation state of the metal exceeds +1. In view of current interest [4, 5] in the nature of the metal-phosphorus bond in transition metal phosphine complexes we have recently synthesised [6, 7] the complexes RuCl<sub>2</sub>(PF<sub>3</sub>)<sub>2</sub>(PPh<sub>3</sub>)<sub>2</sub> (I) and RuCl<sub>2</sub>(PF<sub>3</sub>)(C<sub>10</sub>H<sub>16</sub>) (II) [(C<sub>10</sub>H<sub>16</sub> = 2,7-dimethyl-2,6-diene-1,8-diyl] in which the oxidation state of the ruthenium atom is formally +2 and +4, respectively. We describe below their molecular structures (Figs. 1 and 2) as determined by a single crystal X-ray crystallographic analysis.

I crystallises in space group *P2<sub>1</sub>/n*, a special setting of *P2<sub>1</sub>/c*. The unit cell has dimensions *a* = 18.540(8), *b* = 16.067(7), *c* = 12.415(5) Å and  $\beta$  = 102.64(3)°. A final *R* factor of 3.4% was attained, based on 2310 significant reflections.

II crystallises in space group *C2/c* with unit cell dimensions *a* = 12.575(10), *b* = 9.591(7), *c* = 12.051(9) Å and  $\beta$  = 106.70(6)°. Refinement converged at *R* 3.9%, based on 1112 significant reflections.

In complex I the *cis*-arrangement of the two trifluorophosphine ligands within a slightly distorted octahedral structure confirms our previous suggestion based on NMR studies on the related dihydride complex [6]. The ruthenium-triphenylphosphine distances are found to be 2.471 and 2.456 Å respectively, while the metal-trifluorophosphine bond lengths are very much shorter (2.180 and 2.160 Å). Standard deviations in Ru–P bond lengths are 0.002 Å.

The structure of II is based on a trigonal bipyramid in which the axial posi-

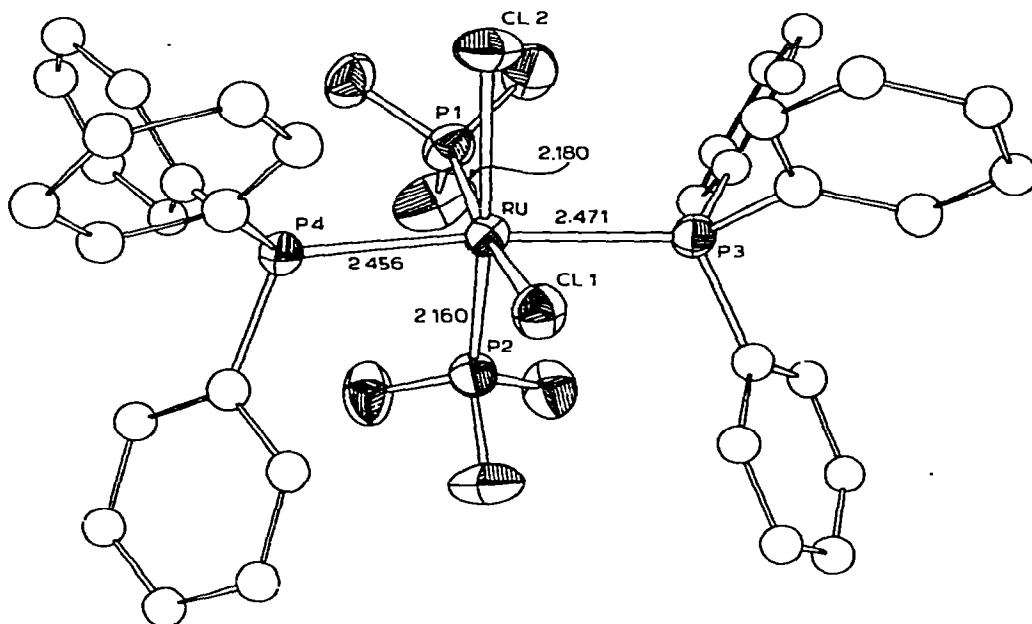


Fig. 1. Molecular structure of the complex  $\text{RuCl}_2(\text{PF}_3)_2(\text{PPh}_3)_2$ .

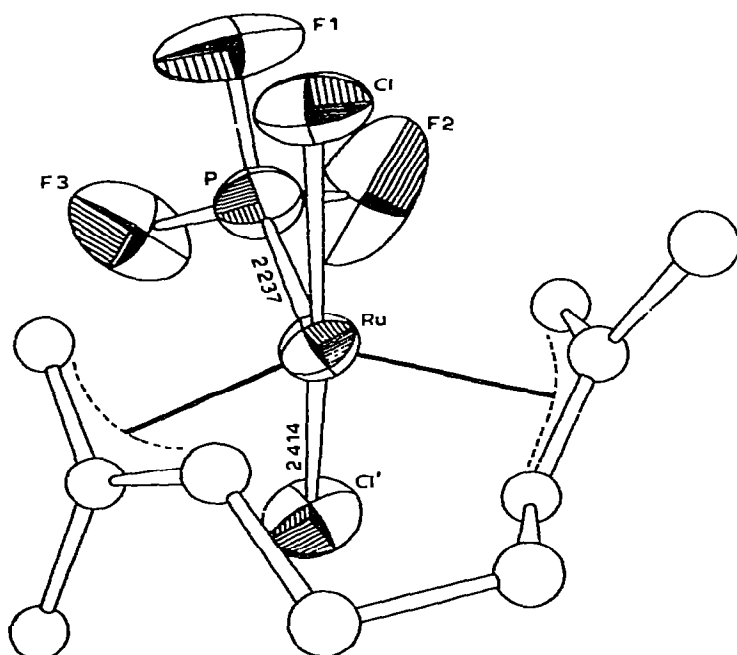


Fig. 2. Molecular structure of the complex  $\text{RuCl}_2(\text{PF}_3)(\text{C}_{10}\text{H}_{16})$ .

tions are occupied by the chlorine atoms. The organic moiety which consists of a linear tail to tail dimer of isoprene occupies two of the equatorial sites and the third position is occupied by  $\text{PF}_3$ . The trifluorophosphine molecule is totally disordered about a two-fold rotation axis along the metal-phosphorus bond.

The ruthenium- $\text{PF}_3$  bond distance in II is found to be 2.237(3) Å which, although significantly longer than the mean of the Ru- $\text{PF}_3$  distances found in I, is nevertheless at the short end of the range of Ru-P distances (2.2-2.4 Å) in other phosphineruthenium complexes [8-18].

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